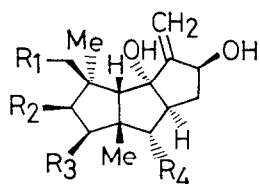


**SYNTHETIC STUDIES ON CAPNELLOL FAMILY :
 AN IMPROVED SYNTHESIS OF $\Delta^{9(12)}$ -CAPNELLENE-3 β ,8 β ,10 α -TRIOL AND
 THE FIRST TOTAL SYNTHESIS OF $\Delta^{9(12)}$ -CAPNELLENE-3 β ,8 β ,10 α ,14-TETROL**

Toshiaki Mase and Masakatsu Shibasaki*

Sagami Chemical Research Center, Nishi-Onnuma, Sagami-hara, Kanagawa 229, Japan

Summary : A much improved synthesis of $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α -triol and the first total synthesis of $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α ,14-tetrol have been accomplished via the common synthetic intermediate 17.



- 1 R₁=R₂=R₃=R₄=H
- 2 R₁=R₃=R₄=H, R₂=OH
- 3 R₁=R₂=R₄=H, R₃=OH
- 4 R₁=R₂=R₃=H, R₄=OH
- 5 R₁=R₃=H, R₂=R₄=OH
- 6 R₂=R₄=H, R₁=R₃=OH

Chart 1

Capnellols are sesquiterpene alcohols 1-6, isolated from sun-dried colonies of the soft coral *Capnella imbricata*.¹ These substances appear to act protective roles against fish predation and invasion by microorganisms, larvae and algae.² Recently we have completed the first total syntheses of (\pm)- $\Delta^{9(12)}$ -capnellene-8 β ,10 α -diol(1) and (\pm)- $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α -triol(3).³ However, the synthetic route to capnellene-triol(3) was quite unsatisfactory in view of the low overall yield so that we continued to study on an improved synthesis of capnellene-triol(3). In this communication we wish to report a much improved synthetic route to capnellene-triol(3) as well as the first total synthesis of the most oxygenated capnellol, (\pm)- $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α ,14-tetrol(6). Both syntheses utilize the common synthetic intermediate 17.

In the previous synthesis³ of capnellene-triol(3), the most unsatisfactory step was the construction of the ABC ring(7 \rightarrow 8) probably owing to the presence of the 1 α -methyl group,⁴ suggesting that later introduction of the 1 α -methyl group would give a more efficient route to 3 (Chart 2).

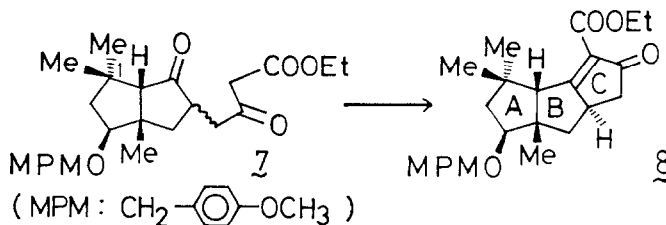
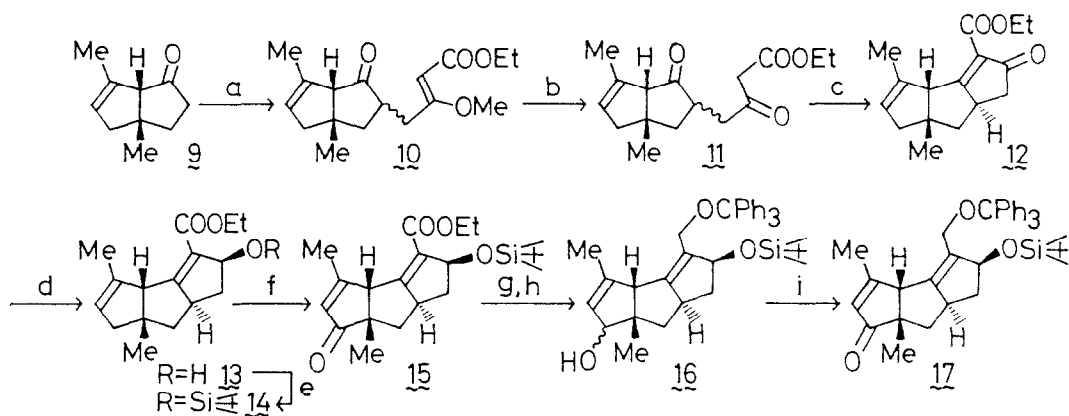


Chart 2

Reaction of the lithium enolate derived from the readily available ketone **9**³ (LDA in THF) with ethyl 4-iodo-3-methoxycrotonate afforded **10**⁵ in 85% yield as an isomeric mixture, which was converted to **11**⁵ in 90% yield on exposure to 30% aqueous perchloric acid. As expected, the β -keto ester **11** was smoothly cyclized to **12**^{5,6} in 85% yield just by treatment with sodium ethoxide in ethanol at room temperature. The tricyclic compound **12** was reduced to the alcohol **13**^{5,7} stereospecifically with NaBH₄-CeCl₃,⁸ followed by protection as TBDMS ether to give **14**⁵ (75% overall yield). Allylic oxidation of **14** with CrO₃-3,5-DMP in CH₂Cl₂⁹ provided the conjugated ketone **15**⁵ (60%). Reduction of **15** with DIBAH and subsequent protection of the primary alcohol with Ph₃CCl-DMAP in DMF gave **16**⁵ (89%), which was further converted to the versatile enone **17**⁵ by treatment with activated MnO₂ in CHCl₃ (94%) (Chart 3).

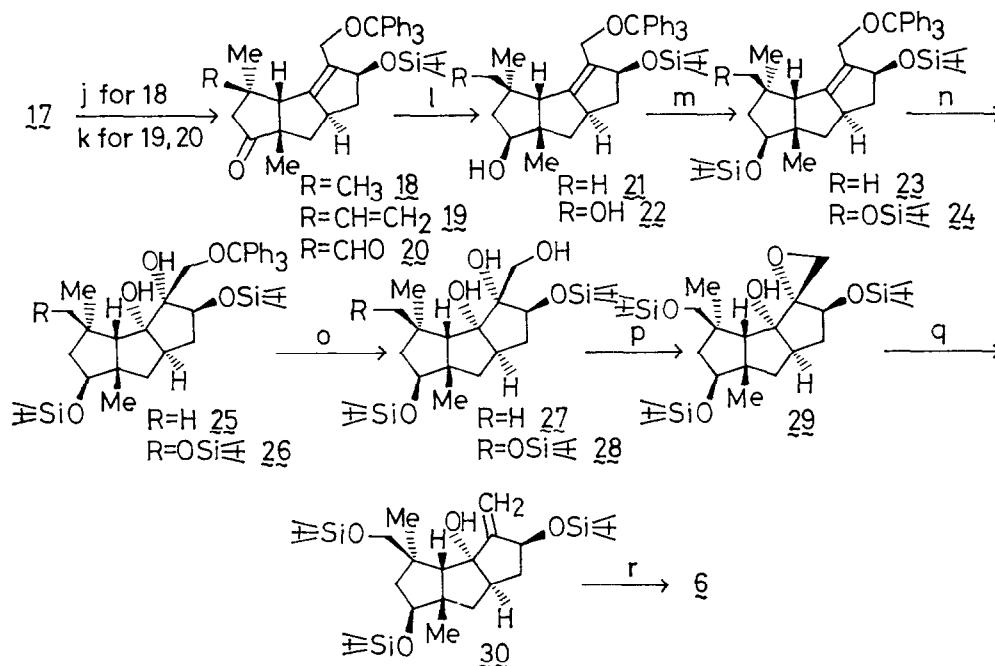


(a) LDA, ICH₂C(OMe)=CHCOOEt, THF, -78°-0°, 85% (b) 30%-HClO₄, Et₂O, r.t. 90% (c) EtONa, EtOH, r.t. 85% (d) NaBH₄, CeCl₃·7H₂O, MeOH, 0°-r.t. 70% (e) TBDMSCl, imidazole, DMF, r.t. 98% (f) CrO₃, 3,5-DMP, CH₂Cl₂, -25°, 60% (g) DIBAH, toluene, -78°-0°, 85% (h) Ph₃CCl, 4-DMAP, DMF, 70°, 95% (i) act. MnO₂, CHCl₃, r.t. 94%

Chart 3

With the key synthetic intermediate **17** in hand, transformation to capnellene-triol(**3**) was first attempted. Conjugate addition of Lipshutz reagent¹⁰ (higher ordered cuprate-BF₃·OEt₂ combination) afforded **18** in 90% yield. Reduction of **18**⁵ with L-Selectride gave **21**,^{5,11} which underwent silylation to furnish **23**⁵ in 81% overall yield. Treatment of **23** with osmium tetroxide in pyridine followed by reductive workup provided the diol **25**⁵ (82%). Chemoselective deprotection of the trityl group in **25** was performed by reaction with Et₂AlCl¹² to afford the known triol **27** (96%), whose spectral data were identical with those of an authentic sample.³ Thus, a much improved total synthesis of capnellene-triol(**3**) has been formally completed (Chart 4).

Using the similar strategy, next we undertook the first total synthesis of capnellene-tetrol(**6**). Conjugate addition of a vinyl group, a representative



(j) $\text{Me}_2\text{Cu}(\text{CN})\text{Li}_2$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, $\text{THF-Et}_2\text{O}$, -50° , 81% (k) $(\text{CH}_2=\text{CH})_2\text{Cu}(\text{CN})\text{Li}_2$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, $\text{THF-Et}_2\text{O}$, -50° , 86%, cat. OsO_4 , NMO, acetone- $\text{H}_2\text{O-tBuOH}$, then, NaIO_4 , dioxane- H_2O , 80% (l) L-Selectride, THF , 0° , 75% for 21, 85% for 22 (m) TBDMSCl , imidazole, DMF , r.t. 99% for 23, 98% for 24 (n) 3 equiv. OsO_4 , pyridine, 40° , then, aq. NaHSO_3 , 70° , 82% for 25, 67% for 26 (o) Et_2AlCl , CH_2Cl_2 , -10° , 96% for 27, 88% for 28 (p) MsCl , NEt_3 , benzene, r.t. then, DBU , r.t. 84% (q) TMSLi , HMPA-THF , -78° -r.t. 70% (r) TBAF , THF , r.t. 96%

Chart 4

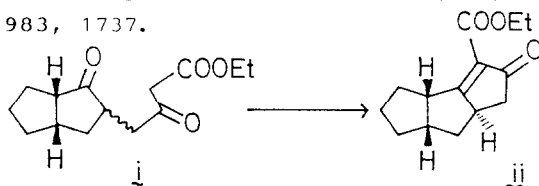
hydroxymethyl equivalent, by using Lipshutz reagent¹⁰ furnished $19^{5,13}$ in 86% yield, which underwent catalytic OsO_4 oxidation¹⁴ and subsequent oxidative cleavage to give the aldehyde 20^5 in 69% overall yield. Reduction of 20 with L-Selectride¹⁵ followed by silylation afforded 24^5 (83% overall yield). After conversion to 26^5 (98%), chemoselective cleavage of the trityl group in 26 was best carried out by treatment with Et_2AlCl to give 28^5 in 88% yield. Reaction of 28 with 1.2 equiv. of $\text{CH}_3\text{SO}_2\text{Cl}$ and 1.2 equiv. of NEt_3 in benzene gave the monomesylate, which was immediately converted to the epoxide 29^5 by treatment with DBU in 84% overall yield. Reaction of 29 with TMSLi in THF-HMPA led to 30^5 (70%), which was followed by exposure to TBAF to provide $(\pm)\text{-}\Delta^9(12)\text{-capnellene-}3\beta,8\beta,10\alpha,14\text{-tetrol}(6)$ (96%) (Chart 4). Spectral data of this synthetic sample were identical with those of the natural product.

In summary, the first total synthesis of $(\pm)\text{-}6$ and a highly efficient synthesis of $(\pm)\text{-}3$ have been accomplished by a general strategy that hopefully will allow the synthesis of other members of the capnellol family. The synthetic routes described above can provide a large amount of $(\pm)\text{-}3$ and $(\pm)\text{-}6$ for detailed biological studies.

Acknowledgement. We thank Professor J.C. Braekman (Université Libre de Bruxelles) for the generous samples of $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α -triol and $\Delta^{9(12)}$ -capnellene-3 β ,8 β ,10 α ,14-tetrol.

References and Notes

- 1) (a) Kaisin, M.; Sheikh, Y.M.; Durham, L.J.; Djerassi, C.; Tursch, B.; Dalozé, D.; Braekman, J.C.; Losman, D.; Karlsson, R. Tetrahedron Lett. **1974**, 2239. (b) Sheikh, Y.M.; Singy, G.; Kaisin, M.; Eggert, H.; Djerassi, C.; Tursch, B.; Dalozé, D.; Braekman, J.C. Tetrahedron **1976**, 32, 1171. (c) Sheikh, Y.M.; Djerassi, C.; Braekman, J.C.; Dalozé, D.; Kaisin, M.; Tursch, B.; Karlsson, R. Tetrahedron **1977**, 33, 2115. (d) Ayanoglu, E.; Gebreyesus, T.; Beechan, C.M.; Djerassi, C.; Kaisin, M. Tetrahedron Lett. **1978**, 1651. (e) Kaisin, M.; Braekman, J.C.; Dalozé, D.; Tursch, B. Tetrahedron **1985**, 41, 1067.
- 2) (a) Ciereszko, L.S. Trans. N.Y. Acad.Sci. **1962**, 24, 502. (b) Burkolder, P.R.; Burkolder, L.M. Science **1958**, 127, 1174. (c) Ciereszko, L.S.; Karns, T.K.B. In "Biology and Geology of Coral Reefs"; Jones, O.A.; Endean, R. Ed.; Academic Press: New York, **1972** ; Vol.2, Chap.6. (e) Tursch, B.; Braekman, J.C.; Dalozé, D.; Kaisin, M. In "Marine Natural Products, Chemical and Biological Perspective" ; Scheuer, P.J., Ed.; Academic Press : New York, **1978** ; Vol.2, Chap.4.
- 3) Shibasaki, M.; Mase, T.; Ikegami, S. J. Am. Chem. Soc. **1986**, 108, 2090. For $\Delta^{9(12)}$ -capnellene-8 α ,10 α -diol, see : Pattenden, G.; Teague, S.J. Tetrahedron Lett. **1982**, 23, 5471.
- 4) The aldol cyclization of the simple β -keto ester **i** afforded the tricyclic compound **ii** in an excellent yield. See: Shibasaki, M.; Mase, T.; Ikegami, S. Chemistry Lett. **1983**, 1737.



- 5) Satisfactory spectroscopic data were obtained for each synthetic intermediate.
- 6) The stereochemistry of **12** was unequivocally determined by the fact that **12** could be efficiently converted to **3** and **6**. See also ref.3.
- 7) The *endo*-alcohol was produced exclusively.
- 8) Gemal, A.L.; Luche, J.-L. J. Am. Chem. Soc. **1981**, 103, 5454.
- 9) Salmond, W.G.; Barta, M.A.; Havens, J.L. J. Org. Chem. **1978**, 43, 2057.
- 10) Lipshutz, B.H.; Parker, D.A.; Kozlowski, J.A.; Nguyen, S.L. Tetrahedron Lett. **1984**, 25, 5959 and references cited therein.
- 11) A ratio of the *exo*- to *endo*-alcohol was 4.5 : 1. The *endo*-alcohol was quantitatively oxidized to **18** for recycling.
- 12) Köster, H.; Sinha, N.D. Tetrahedron Lett. **1982**, 23, 2641.
- 13) The 1 β -vinyl isomer was produced exclusively.
- 14) The double bond at C₉-C₁₀ remained intact under these reaction conditions.
- 15) A ratio of the *exo*- to *endo*-alcohol was 8 : 1.

(Received in Japan 19 July 1986)